



Mechanisms of gas migration in flooding post-mining context

Experimental approach and modelling



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- > Since 1990
- > Budget : 70 M€
- > 600 employees

3 main divisions :

- Accidental Risks Division
- Chronic Risks Division
- Ground and underground Risks Division

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Context

European project FLOMINET
Support mission to state services: abandoned coal mines of France
Thesis objectives

- To characterize mine gas migration (CH₄)
- Flooding impact on gas sorption/release

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Context

CASPER experiment
Sorption under hydrostatic pressure
Solubilization

Slide 4

Context

HYTEC
Reactive transport code

CHESS
Chemical equilibrium code

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Context

HYTEC Development
Two-phase flow

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Laboratory experiment

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CASPER : objectives
 (sorption capacity under high hydrostatic pressure during rock flooding)

To reproduce flooding conditions
 To measure released gas from coal for given hydrostatic pressure

CH₄ saturated coal

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CASPER : objectives
 (sorption capacity under high hydrostatic pressure during rock flooding)

To reproduce flooding conditions
 To measure released gas from coal for given hydrostatic pressure
 To determine which pressure stops gas release

CH₄(aq) release

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CASPER: initial system

Water pressure: stop gas release, H₂O/CH₄ competition on sites ?

Hypotheses:

- Presence of water decreases CH₄ sorption (e.g. Joubert et al., 1973 ; Crosdale et al., 2008 ; Charrière, 2009)
- Released CH₄ enters aqueous phase, no gas in the cell

First result (quick and dirty)

- Experiment released amount : **0.003 mol** (water degassing)
- Validation with solubility model of Duan and Mao (2006)

$$\ln \frac{V_{CH_4}^g}{m_{CH_4}} = \frac{\mu_{CH_4}^g}{RT} - \ln \phi_{CH_4} + \sum_c 2\lambda_{CH_4-c} m_c + \sum_a 2\lambda_{CH_4-a} m_a + \sum_i \sum_j \tilde{\epsilon}_{CH_4-c-i} m_c m_i$$

Labels: Gas molar fraction, Chemical potential, Fugacity coefficient, CH₄ - cations - anions interactions, Molality

- Soluble amount in the cell : **0.0036 mol**

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CASPER

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CASPER: sampling/analysis system

Water with aqueous gas sampling: ROLSI™ (Rapid On Line Sampling Injector)
 Analysis : gas chromatograph
 CH₄(aq) monitoring

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Experimental result fit

Reproduction of experimental conditions

Defined mineral species ↔ coal

- Sorption sites

$$\text{CoalSite} + \text{CH}_4(\text{aq}) = \text{CoalSite-CH}_4$$

Adsorption constant K

$$K = \frac{[\text{CoalSite} - \text{CH}_4]}{[\text{CoalSite}][\text{CH}_4(\text{aq})]} = \frac{[N_{\text{TOT sites}}] - [\text{CH}_4(\text{aq})]}{[\text{CH}_4(\text{aq})]^2}$$

- Log (K) = 1.87
- (P = 25.5 bar, T = 26°C)

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Modelling

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HYTEC
(Hydrological Transport coupled with Equilibrium Chemistry)

Reactive transport

Current code

- Flow, advection, diffusion, dispersion
- Mobile and immobile phases
- Chemical reactions : CHES

CHES (Chemical Equilibrium of Species and Surfaces)

- Speciation
- Adsorption
- Chemical equilibrium
- Kinetics

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CH₄(aq) migration/adsorption

CH₄(aq) plume migration

- Gallery K = 10⁻¹ m/s (white)
- Fractured coalbed K = 10⁻⁴ m/s (blue)
- Fresh rock K = 10⁻⁶ m/s (green)
- Head gradient Δh = 0.01

Impact of fracture

- Fracture K = 10⁻⁴ m/s (blue)

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CH₄(aq) migration/adsorption

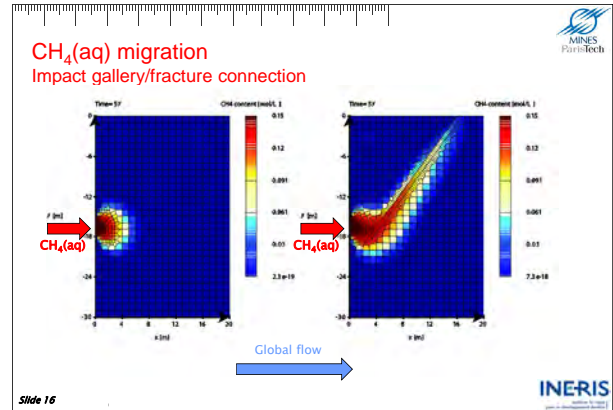
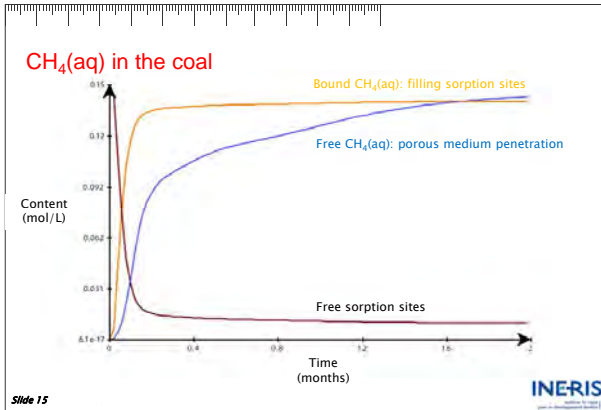
CH₄(aq) plume migration

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CH₄(aq) migration/adsorption

CH₄(aq) plume migration

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Wrapping up & prospects

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Wrapping up

Several phenomena to take into account

CASPER

- First result consistent with bibliography
- New protocol operational end of summer 2010 : faster, better quality

HYTEC

- Sorption equilibrium constant
- CH₄(aq) migration in different structures

Prospects

Flooding modelling in development (unsaturated flow)

Two-phase flow : end of the year 2010

Development of tools for gas risks management in flooded mines

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Thank you for attention

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